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Ortenauhalle Congress 1
Deep geothermal energy



Viscosified System for Enhanced Acidizing of Geothermal Wells in Sandstone Formations

Tensid-basiertes System zur verbesserten Säurebehandlung von Geothermiebohrungen in Sandsteinformationen

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In geothermal projects, the main goal of acidizing injector wells in sandstones is scale removal and consequently injectivity increase. At low temperature, hydrochloric-based treatment fluids are commonly employed for this purpose. In this context, employing diverting agents to prevent acids from leaking into the most permeable sub-layer of the target zone is recommendable.

This paper presents the surfactant-based product SDA-550 which shows a tendency of forming rodlike micelles in acidic solutions. As shown in Figure 1, a chaotic worm-like arrangement of dissolved molecules leads to an increase in viscosity. This behavior creates a temporary blocking effect which causes fluid diversion and facilitates successful acidizing.

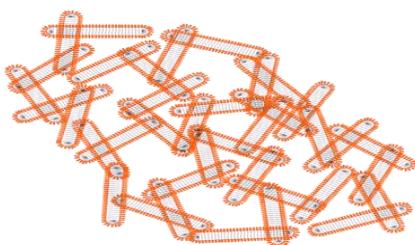


Figure 1: Macro-structure of surfactant-based product SDA-550 in acidic solutions.

Preparing the first field trial in a Dutch geothermal well, we performed extensive lab experiments regarding solubility of solid samples, corrosion of metal coupons, and rheology of acidic recipes containing different concentrations of SDA-550. The following table sums up the results of solubility testing with actual bailer samples from the well:

| Fluid | Composition | | |
|-----------------------|--|----------------------|------------------|
| 1 | 15% Hydrochloric Acid | | |
| 2 | 15% Hydrochloric Acid + 25 kg/m ³ SCC-240 | | |
| Test Parameter | | | |
| Temperature, °C | Δp N ₂ , psi | Exposure Time, hours | Fluid Volume, mL |
| 40 | Atm. | 2 | 50 |
| Test Results | | | |
| Weight before, g | Weight after, g | Weight Loss, g | Solubility, % |
| 1.003 | 0.498 | 0.505 | 50 |
| 1.004 | 0.277 | 0.727 | 72 |

As shown in Figure 2, the dosage of corrosion inhibitor added to the acidic systems suffices to protect L-80 metal coupons.

| Fluid | Composition | | |
|---|---|---|--|
| 1 | 15% Hydrochloric Acid + 10 L/m ³ SCI-123 | | |
| 2 | 15% Hydrochloric Acid + 25 kg/m ³ SCC-240 + 10 L/m ³ SCI-123 | | |
| Test Parameter | | | |
| Temperature, °C | Δp N ₂ , psi | Test Time @ 75°C, hours | Test Time @ 40°C, hours |
| 75 / 40 | 1,000 | 7 | 14 |
| After 7 hours @ 75°C in Fluid 1 | | | |
| Weight Loss, % | Material Loss, lbs./ft ² | Weight Loss, % | Material Loss, lbs./ft ² |
| 0.028 | 0.001 | 0.032 | 0.001 |
| Coupon A: Before | Coupon A: After | Coupon B: Before | Coupon B: After |
|  |  |  |  |

Figure 2: Results of corrosion testing with two different acidic recipes and L-80 coupons, as determined after 7 hours at 75°C.

For dissolving carbonates, as well as silicates in the Slochteren sandstone formation (BHST at approximately 75°C), HCl- and HCl/HF-based recipes were pumped in a stepwise approach (see table below). The addition of SDA-550 and hence the viscosity of the diverter step was adjusted based upon lab experience and in accordance with the client.

| Step | Volume |
|--|--------------------|
| 1 15% Hydrochloric Acid | 10 m ³ |
| 2 5% Hydrochloric Acid + SDA-550 (viscosified Fluid) | 4 m ³ |
| 3 15% Hydrochloric Acid | 6 m ³ |
| 4 15% Hydrochloric Acid + 25 kg/m ³ SCC-240 | 15 m ³ |
| 5 5% Hydrochloric Acid | 5 m ³ |
| 6 Displacement 1 with Formation Water | 91 m ³ |
| 7 Displacement 2 with Formation Water | 20 m ³ |
| 8 Displacement 3 with Formation Water | 15 m ³ |
| 9 Reaction Time | 0 m ³ |
| 10 Injection Test | 198 m ³ |

The subsequent injection test with brine revealed a significant improvement in injectivity of the formation. While keeping the well head pressure at a constant level, we could increase the pumping rate by a factor of four.

Figure 3 summarizes the pumping schedule and Figure 4 shows equipment on site.

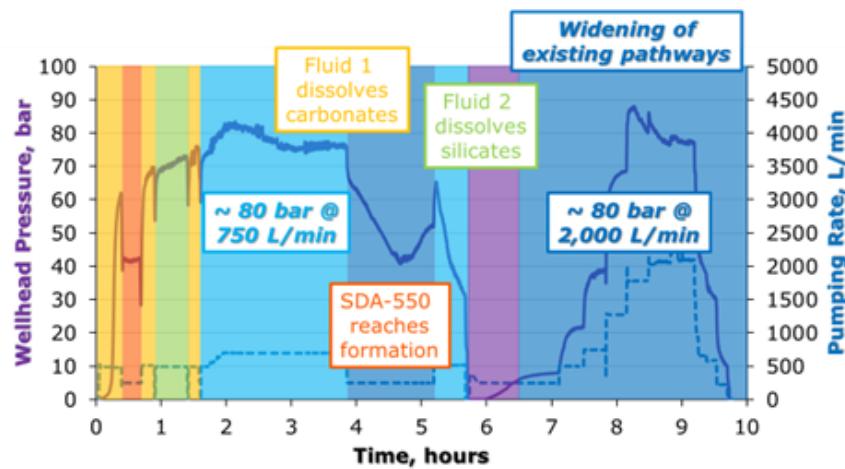


Figure 3: Pumping schedule.



Figure 4: Equipment on site.

This great result was achieved due to the superior chemical properties of our innovative diverter agent combined with the great effectiveness of the tailor-made treatment fluids. Thus, laboratory and field results impressively proof that we have reached the next level of acidizing sandstones..